# organic compounds

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# 4-Chloro-N-cyclohexylbenzenesulfonamide

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Key indicators: single-crystal X-ray study; T = 296 K; mean  $\sigma$ (C–C) = 0.005 Å; R factor = 0.053; wR factor = 0.150; data-to-parameter ratio = 21.4.

The title compound, C<sub>12</sub>H<sub>16</sub>ClNO<sub>2</sub>S, adopts an L-shaped conformation, with the central C-S-N-C torsion angle being  $-78.0 (2)^{\circ}$ . The cyclohexyl ring adopts a chair conformation. In the crystal, adjacent molecules are connected by pairs of N-H···O hydrogen bonds around an inversion centre, forming cyclic dimers [graph set  $R_2^2(8)$ ].

#### **Related literature**

For background to the pharmacological uses of sulfonamides, see: Korolkovas (1988); Mandell & Sande (1992). For related structures, see: Sharif et al. (2011); Khan et al. (2010); John et al. (2010). For ring conformational analysis, see: Cremer & Pople (1975).



#### **Experimental**

Crystal data

C12H16CINO2S  $M_r = 273.78$ Monoclinic,  $P2_1/c$ a = 11.1226 (5) Å b = 6.2490 (2) Å c = 19.8635 (9) Å  $\beta = 96.505 \ (2)^{\circ}$ 

$V = 1371.73 (10) \text{ Å}^3$
Z = 4
Mo $K\alpha$ radiation
$\mu = 0.42 \text{ mm}^{-1}$
T = 296  K
$0.29 \times 0.15 \times 0.11 \ \mathrm{mm}$

Data collection

1 restraint

B and concernon	
Bruker APEXII CCD diffractometer 12714 measured reflections	3365 independent reflections 2075 reflections with $I > 2\sigma(I)$ $R_{int} = 0.030$
Refinement	
$R[F^2 > 2\sigma(F^2)] = 0.053$	H atoms treated by a mixture of
$wR(F^2) = 0.150$	independent and constrained
S = 1.03	refinement
3365 reflections	$\Delta \rho_{\rm max} = 0.39 \ {\rm e} \ {\rm \AA}^{-3}$
157 parameters	$\Delta \rho_{\rm min} = -0.25 \text{ e} \text{ Å}^{-3}$

#### Table 1 Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N1 - H1N \cdots O1^{i}$	0.85 (4)	2.05 (4)	2.891 (4)	170 (3)

Symmetry code: (i) -x + 1,  $y + \frac{1}{2}$ ,  $-z + \frac{1}{2}$ .

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SIR97 (Altomare et al., 1999); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999) and PLATON (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG5164).

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supplementary materials

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### 4-Chloro-N-cyclohexylbenzenesulfonamide

## S. Sharif, S. Y. Mughal, I. U. Khan, M. Akkurt and M. H. Khan

#### Comment

Sulfonamide group containing drugs are extensively used for the treatment of certain infections caused by Gram-positive and Gram-negative microorganisms (Korolkovas, 1988; Mandell & Sande, 1992). In continuation of our on going structural studies of cyclohexylamine and sulfonamides synthesis (John *et al.*, 2010; Khan *et al.*, 2010; Sharif *et al.*, 2011), herein the crystal structure of title compound (I) is described.

In (I), (Fig. 1), the S atom has a distorted tetrahedral geometry within a CNO<sub>2</sub> donor set [maximum deviation: O—S—O = 119.45 (12)°]. The central C6–S1–N1–C7 torsion angle is -78.0 (2)°. The C7–C12 cyclohexyl ring adopts a chair conformation, with puckering parameters (Cremer & Pople, 1975) Q = 0.536 (4) Å,  $\theta$  = 180.0 (4) °,  $\varphi$  = 196 (16) °.

In the crystal, two adjacent molecules are linked by a pair of N—H···O hydrogen bonds, forming an inversion dimer with an  $R_2^2(8)$  ring motif (Table 1, Fig. 2).

#### **Experimental**

To 115  $\mu$ l (1 mmol) of cyclohexylamine in 10 ml distilled water, was added 211 mg (1 mmol) of 4-chlorobenzenesulfonyl chloride while maintaining the pH of reaction mixture at 8 by using 3% sodium carbonate solution. Consumption of the reactants was confirmed by TLC. The pH of reaction mixture was adjusted by 3 N HCl at 3. Precipitates formed, washed with water and crystallized from methanol.

#### Refinement

The NH H–atom was located in a difference Fourier map and isotropically refined with a distance restraint: N—H = 0.86 (2) Å. C-bound H atoms were placed in calculated positions with C—H distances in the range 0.93–0.98 Å and and were refined using a riding model with  $U_{iso}(H) = 1.2U_{eq}(C)$ . In the final refinement two low angle reflections evidently effected by the beam stop were omitted, *i.e.* 0 0 2 and 1 0 0.

#### Figures



Fig. 1. The title molecule, showing the atom numbering scheme. Displacement ellipsoids for non-H atoms are drawn at the 30% probability level.



Fig. 2. The crystal packing of (I) down *b* axis, showing the molecules are linked into dimers by pairs of N—H  $\cdots$  O hydrogen bonds, forming  $R_2^2(8)$  graph-set motif. Hydrogen atoms not involved in hydrogen bonding have been omitted for clarity.

F(000) = 576 $D_{\rm x} = 1.326 \text{ Mg m}^{-3}$ 

 $\theta = 2.6-21.6^{\circ}$   $\mu = 0.42 \text{ mm}^{-1}$ T = 296 K

Needle, light brown  $0.29 \times 0.15 \times 0.11 \text{ mm}$ 

Mo K $\alpha$  radiation,  $\lambda = 0.71069$  Å Cell parameters from 3062 reflections

#### 4-Chloro-N-cyclohexylbenzenesulfonamide

C <sub>12</sub> H <sub>16</sub> ClNO <sub>2</sub> S
$M_r = 273.78$
Monoclinic, $P2_1/c$
Hall symbol: -P 2ybc
<i>a</i> = 11.1226 (5) Å
<i>b</i> = 6.2490 (2) Å
c = 19.8635 (9)  Å
$\beta = 96.505 \ (2)^{\circ}$
$V = 1371.73 (10) \text{ Å}^3$
Z = 4

#### Data collection

Bruker APEXII CCD diffractometer	2075 reflections with $I > 2\sigma(I)$
Radiation source: sealed tube	$R_{\rm int} = 0.030$
graphite	$\theta_{\text{max}} = 28.3^{\circ}, \ \theta_{\text{min}} = 3.4^{\circ}$
$\varphi$ and $\omega$ scans	$h = -14 \rightarrow 14$
12714 measured reflections	$k = -6 \rightarrow 8$
3365 independent reflections	$l = -26 \rightarrow 26$

#### Refinement

Refinement on  $F^2$ Least-squares matrix: full  $R[F^2 > 2\sigma(F^2)] = 0.053$  $wR(F^2) = 0.150$ S = 1.033365 reflections 157 parameters 1 restraint Primary atom site location: structure-invariant direct methods Secondary atom site location: difference Fourier map Hydrogen site location: inferred from neighbouring sites H atoms treated by a mixture of independent and constrained refinement  $w = 1/[\sigma^2(F_o^2) + (0.0613P)^2 + 0.526P]$ where  $P = (F_o^2 + 2F_c^2)/3$ (Δ/σ)<sub>max</sub> < 0.001 Δρ<sub>max</sub> = 0.39 e Å<sup>-3</sup> Δρ<sub>min</sub> = -0.25 e Å<sup>-3</sup>

#### Special details

**Geometry**. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

**Refinement**. Refinement on  $F^2$  for ALL reflections except those flagged by the user for potential systematic errors. Weighted *R*-factors *wR* and all goodnesses of fit *S* are based on  $F^2$ , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative  $F^2$ . The observed criterion of  $F^2 > \sigma(F^2)$  is used only for calculating *-R*-factor-obs *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on  $F^2$  are statistically about twice as large as those based on *F*, and *R*-factors based on ALL data will be even larger.

	x	У	Ζ	$U_{\rm iso}*/U_{\rm eq}$
Cl1	0.97047 (9)	0.63719 (19)	0.09885 (6)	0.1159 (5)
S1	0.67445 (5)	0.02579 (10)	0.26455 (3)	0.0538 (2)
01	0.57744 (18)	-0.0412 (3)	0.21563 (11)	0.0788 (8)
O2	0.75531 (17)	-0.1301 (3)	0.29647 (11)	0.0710 (7)
N1	0.61305 (18)	0.1518 (3)	0.32146 (11)	0.0527 (7)
C1	0.7127 (2)	0.3981 (4)	0.19816 (13)	0.0581 (9)
C2	0.7777 (3)	0.5312 (4)	0.16141 (14)	0.0650 (9)
C3	0.8901 (3)	0.4698 (5)	0.14686 (14)	0.0654 (10)
C4	0.9396 (2)	0.2788 (5)	0.16961 (16)	0.0733 (11)
C5	0.8750 (2)	0.1454 (4)	0.20712 (14)	0.0606 (9)
C6	0.7611 (2)	0.2037 (4)	0.22098 (12)	0.0465 (7)
C7	0.6789 (2)	0.2145 (4)	0.38711 (12)	0.0541 (8)
C8	0.7329 (3)	0.4325 (5)	0.38666 (16)	0.0829 (11)
C9	0.7924 (4)	0.4979 (7)	0.45609 (18)	0.1069 (17)
C10	0.7084 (4)	0.4802 (9)	0.5084 (2)	0.117 (2)
C11	0.6528 (5)	0.2674 (10)	0.50951 (18)	0.135 (2)
C12	0.5920 (4)	0.1970 (7)	0.44002 (17)	0.1033 (16)
H1	0.63600	0.43810	0.20780	0.0700*
H1N	0.555 (3)	0.234 (6)	0.3060 (19)	0.1390*
H2	0.74580	0.66260	0.14640	0.0780*
H4	1.01630	0.23970	0.15970	0.0880*
Н5	0.90830	0.01600	0.22310	0.0730*
H7	0.74450	0.11150	0.39850	0.0650*
H8A	0.67010	0.53500	0.37150	0.0990*
H8B	0.79270	0.43550	0.35470	0.0990*
H9A	0.86230	0.40740	0.46850	0.1280*
H9B	0.82060	0.64450	0.45420	0.1280*
H10A	0.75200	0.51000	0.55240	0.1410*
H10B	0.64520	0.58700	0.49970	0.1410*
H11A	0.59300	0.26840	0.54140	0.1620*
H11B	0.71460	0.16380	0.52540	0.1620*
H12A	0.56470	0.05020	0.44260	0.1240*

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(A^2)$ 

# supplementary materials

H12B	0.52180	0.28630	0.42710	0.12	240*			
Atomic displacement parameters $(Å^2)$								
	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	U <sup>23</sup>		
C11	0.0867 (7)	0.1367 (9)	0.1293 (9)	-0.0197 (6)	0.0339 (6)	0.0505 (7)		
S1	0.0456 (4)	0.0438 (3)	0.0725 (4)	-0.0069 (3)	0.0087 (3)	-0.0065 (3)		
01	0.0634 (12)	0.0815 (13)	0.0904 (14)	-0.0310 (10)	0.0034 (11)	-0.0211 (11)		
O2	0.0691 (12)	0.0436 (9)	0.1030 (15)	0.0083 (9)	0.0210 (11)	0.0082 (9)		
N1	0.0393 (11)	0.0562 (12)	0.0629 (13)	0.0041 (9)	0.0071 (10)	0.0035 (10)		
C1	0.0540 (15)	0.0589 (15)	0.0637 (16)	0.0058 (13)	0.0169 (13)	0.0006 (12)		
C2	0.0704 (18)	0.0594 (15)	0.0664 (16)	0.0034 (14)	0.0132 (15)	0.0056 (13)		
C3	0.0553 (16)	0.0765 (18)	0.0646 (16)	-0.0132 (15)	0.0076 (13)	0.0050 (14)		
C4	0.0398 (14)	0.091 (2)	0.090 (2)	-0.0036 (15)	0.0118 (15)	0.0011 (18)		
C5	0.0418 (14)	0.0602 (15)	0.0793 (18)	0.0021 (12)	0.0048 (13)	0.0025 (13)		
C6	0.0410 (12)	0.0461 (12)	0.0514 (13)	-0.0033 (10)	0.0015 (10)	-0.0093 (10)		
C7	0.0463 (14)	0.0565 (14)	0.0597 (15)	0.0112 (12)	0.0063 (12)	0.0059 (12)		
C8	0.095 (2)	0.082 (2)	0.0727 (19)	-0.0204 (19)	0.0134 (18)	-0.0077 (16)		
C9	0.106 (3)	0.119 (3)	0.095 (3)	-0.014 (2)	0.009 (2)	-0.038 (2)		
C10	0.092 (3)	0.178 (5)	0.082 (2)	0.019 (3)	0.009 (2)	-0.049 (3)		
C11	0.132 (4)	0.216 (6)	0.059 (2)	-0.014 (4)	0.018 (2)	0.022 (3)		
C12	0.100 (3)	0.141 (3)	0.072 (2)	-0.027 (2)	0.023 (2)	0.021 (2)		

## Geometric parameters (Å, °)

Cl1—C3	1.731 (4)	C10-C11	1.468 (8)
S1—O1	1.431 (2)	C11—C12	1.531 (6)
S1—O2	1.425 (2)	C1—H1	0.9300
S1—N1	1.594 (2)	С2—Н2	0.9300
S1—C6	1.762 (3)	C4—H4	0.9300
N1—C7	1.475 (3)	С5—Н5	0.9300
N1—H1N	0.85 (4)	С7—Н7	0.9800
C1—C2	1.366 (4)	C8—H8A	0.9700
C1—C6	1.384 (4)	C8—H8B	0.9700
C2—C3	1.370 (5)	С9—Н9А	0.9700
C3—C4	1.370 (4)	С9—Н9В	0.9700
C4—C5	1.374 (4)	C10—H10A	0.9700
C5—C6	1.376 (3)	C10—H10B	0.9700
C7—C12	1.510 (5)	C11—H11A	0.9700
С7—С8	1.489 (4)	C11—H11B	0.9700
C8—C9	1.517 (5)	C12—H12A	0.9700
C9—C10	1.478 (6)	C12—H12B	0.9700
O1—S1—O2	119.45 (12)	C5—C4—H4	120.00
O1—S1—N1	106.00 (12)	С4—С5—Н5	120.00
O1—S1—C6	105.28 (12)	С6—С5—Н5	120.00
O2—S1—N1	108.77 (12)	N1—C7—H7	108.00
O2—S1—C6	107.29 (11)	С8—С7—Н7	108.00
N1—S1—C6	109.82 (11)	С12—С7—Н7	108.00

S1—N1—H1N	114 (3)		C7—C8—H8B		109.00
C7—N1—H1N	116(3)		C9—C8—H8A		109.00
$C^2 - C^1 - C^6$	1200(2)		C9—C8—H8B		109.00
$C_1 - C_2 - C_3$	120.0(2) 119.6(3)		H8A_C8_H8B		108.00
C1 - C2 - C3	119.2 (2)		C8—C9—H9A		109.00
$C_2 - C_3 - C_4$	117.2(2) 1211(3)		C8—C9—H9B		109.00
Cl1-C3-C4	121.1(3) 1197(2)		C10-C9-H9A		109.00
$C_{3}$ $C_{4}$ $C_{5}$	119.7(2) 119.5(2)		C10—C9—H9B		109.00
C4—C5—C6	119.9 (2)		H9A—C9—H9B		108.00
S1—C6—C1	120.05 (17)		C9—C10—H10A		109.00
S1—C6—C5	119.92 (19)		C9—C10—H10B		109.00
C1—C6—C5	120.0 (2)		C11—C10—H10A		109.00
N1—C7—C8	113.5 (2)		C11—C10—H10B		109.00
C8—C7—C12	111.2 (3)		H10A—C10—H10B		108.00
N1—C7—C12	107.7 (2)		C10-C11-H11A		109.00
C7—C8—C9	112.1 (3)		C10-C11-H11B		109.00
C8—C9—C10	111.9 (4)		C12—C11—H11A		109.00
C9—C10—C11	112.3 (4)		C12—C11—H11B		109.00
C10-C11-C12	113.0 (4)		H11A—C11—H11B		108.00
C7—C12—C11	110.8 (4)		C7—C12—H12A		110.00
C2—C1—H1	120.00		C7—C12—H12B		109.00
С6—С1—Н1	120.00		C11—C12—H12A		109.00
С1—С2—Н2	120.00		C11—C12—H12B		109.00
С3—С2—Н2	120.00		H12A—C12—H12B		108.00
С3—С4—Н4	120.00				
01—S1—N1—C7	168.78 (18)		C1—C2—C3—Cl1		178.7 (2)
O2—S1—N1—C7	39.2 (2)		Cl1—C3—C4—C5		-179.3 (2)
C6—S1—N1—C7	-78.0 (2)		C2—C3—C4—C5		0.6 (5)
N1—S1—C6—C5	135.6 (2)		C3—C4—C5—C6		0.6 (4)
O2—S1—C6—C1	-165.3 (2)		C4—C5—C6—S1		176.1 (2)
N1—S1—C6—C1	-47.2 (2)		C4—C5—C6—C1		-1.1 (4)
O1—S1—C6—C1	66.5 (2)		N1-C7-C8-C9		176.1 (3)
O2—S1—C6—C5	17.5 (2)		С12—С7—С8—С9		54.6 (4)
O1—S1—C6—C5	-110.7 (2)		N1-C7-C12-C11		-178.0 (3)
S1—N1—C7—C12	-144.8 (2)		C8—C7—C12—C11		-53.2 (4)
S1—N1—C7—C8	91.8 (2)		С7—С8—С9—С10		-54.4 (4)
C6—C1—C2—C3	0.7 (4)		C8—C9—C10—C11		53.3 (5)
C2-C1-C6-S1	-176.7 (2)		C9—C10—C11—C12		-53.2 (6)
C2-C1-C6-C5	0.5 (4)		C10-C11-C12-C7		53.0 (5)
C1—C2—C3—C4	-1.3 (4)				
Hydrogen-bond geometry (Å, °)					
D—H···A		<i>D</i> —Н	H···A	$D \cdots A$	D—H··· $A$
N1—H1N····O1 <sup>i</sup>		0.85 (4)	2.05 (4)	2.891 (4)	170 (3)

N1—H1N···O1<sup>i</sup> Symmetry codes: (i) -x+1, y+1/2, -z+1/2.







